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(Rodd's Chemistry Of Carbon Compounds. 2nd  
Edition) By Ernest H. Rodd**

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hydrocarbon, carbon: hydrocarbons any of a class of organic chemical compounds composed only of the have molecular formulas that satisfy the general expression

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as is indicated in the following general Condensed Structural Formula Melting the double bond has five carbon atoms, so the compound is a

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Why are some carbon compounds classified as inorganic? From a database of frequently asked questions from the Introduction to inorganic chemistry section of General

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The study of the structure, preparation, properties, and reactions of carbon compounds. Organic Chemistry . (Journal of General Chemistry; since 1931),

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General properties; Name, symbol: the large majority of all chemical compounds Carbon also has the highest sublimation point of all elements. At

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A chemical compound If the formula contains no carbon, Kenneth W.; Davis, Raymond E.; Peck, M. Larry (2000), General Chemistry (6th ed.),

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Rodd's Chemistry of Carbon Compounds. Herausgegeben von S. Coffey. Vol. I, Part A: General Introduction, Hydrocarbons, Halogen Derivatives. Elsevier Publishing Comp

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Answer: Organic chemistry is the study of carbon compounds of living matter i.e., For e.g., for alkanes the general formula is  $C_n H_{2n+2}$ ,

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Correct Answer Number: 1. Explanation: In general, organic molecules contain carbon and organic chemistry is the study of carbon compounds. More Information

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Rodd's Chemistry of Carbon Compounds Vol. I A: General Introduction, Hydrocarbons; Halogen Derivatives. S. Coffey. 569S., zahlr. Abb. und Tab. 1964, Elsevier

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