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March 31-April 1, 1970. proceedings of the International Symposium on University Chemical Education held in Electronic spectra and quantum chemistry.

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Nov 09, 2007 I believe an anti-aromatic compound is one that should, from its p-orbitals be as stable as benzene, but is actually not as stable as it would be

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To find whether a ring is aromatic, antiaromatic, nonaromatic, or none of the above, use the following flow chart. see also

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Proceedings of 9th International Conference on In: International Journal of Organic Chemistry. 1 anti-Zeno effect, quantum

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Aromaticity Aromaticity is structural distortions and formation of islands of aromaticity in the system, which is now globally anti-aromatic. First-order

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If an anti-aromatic ring can become non aromatic by conforming to nonplanar, it will. The NH, O, and S you refer to are probably pyrrole, furan, and thiphen.

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ring would be anti-aromatic, and therefore highly unstable, due to the symmetry of the combinations of p atomic orbitals. By twisting the ring,

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Aromaticity is a property of conjugated cycloalkenes in which the stabilization of the molecule is strong due to the ability of the electrons in the  $\pi$

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Journal of the Chemical Society D: Chemical Communications was published between 1969 - 1971.

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Antiaromaticity is a characteristic of a cyclic molecule with a electron system that has higher energy due to the presence of  $4n$  electrons in it. Unlike aromatic

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Quasi-aromaticity what does it mean? (Eds.), Aromaticity, Pseudo-aromaticity, Anti-aromaticity, Proceedings of an International Symposium; Jerusalem,

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Lloyd, D. M.: Proceedings. Contribution to the International Symposium on Aromatic, Pseudo-Aromaticity, Anti-Aromaticity, Jerusalem 1970, p. 85.

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